

SUPPORTING INFORMATION: COMPARISON OF ACTIVATION METHODS OF 3D-PRINTED ELECTRODES FOR MICROBIAL ELECTROCHEMICAL TECHNOLOGIES

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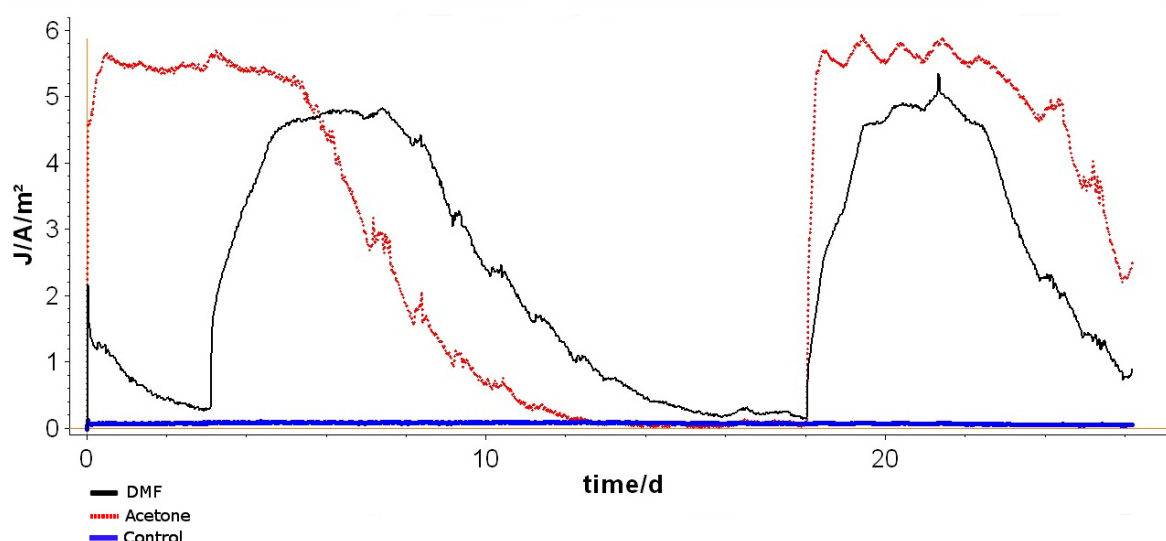


Figure S1. Current density profiles of two successive cycles at the end of a 60 days period that allowed for the development of a stable biofilm.

Electrochemical Impedance Spectroscopy Analysis (EIS)

The EIS spectra obtained during the abiotic characterization of the PLA/Graphene electrodes (Figure 3) were fitted to equivalent circuits to facilitate the interpretation of the results. Two equivalent circuits (EC) have been proposed. Circuit 1 (Fig. S11 1)) has proved to be useful in the modelling of successfully activated electrodes (DMF and acetone). The main characteristics of the spectra can be explained using a modified pore model version. In this model the impedance of the pore wall is modelled by a charge transfer resistance (R_2), a Warburg impedance (W_2) and a constant phase element (Q_2) accounts the pseudo-capacitive processes that occur at the electrode-electrolyte interface. Table 1 summarizes the fitted parameters to EC 1. The parameter a_2 allows to compare the near capacitive behavior of DMF electrode in contrast to the pseudo-capacitive behaviour of acetone electrode ($a_2=1$ in an ideal capacitor).

EC 2 has served as model from control and electrochemical-treated electrodes. Its elements respond to the modelling of two different pseudo-capacitive processes, perhaps because they occur at two different spatial scales.

The impedances of constants phase elements were adjusted to the expression:

$$Z_Q = \frac{1}{Q\omega(j\omega)^a}$$

Where Q is the pseudocapacitance (F.s^{a-1}), j the imaginary unit, ω is the angular frequency (s⁻¹) and a is an empirical dimensionless parameter (0 for a resistor and 1 for an ideal capacitor).

The impedance of Warburg element is described by:

$$Z_W = \frac{S}{\sqrt{j\omega}}$$

Where S is in (Ω .s^{-1/2}), j the imaginary unit, ω is the angular frequency (s⁻¹).

Electrode	Equivalent circuit	R ₁ (Ω)	R ₂ (Ω)	Q ₂ (F.s ^{a-1})	a ₂	S ₂ (Ω .s ^{-1/2})
DMF	(1)	14.5	313.8	5.38x10 ⁻⁴	0.907	23.38
Acetone	(1)	23.6	238	1.04x10 ⁻⁴	0.623	9.81

Table S1. Fitted parameters to EC 1 (DMF and acetone electrodes).

Electrode	Equivalent circuit	R ₁ (Ω)	R ₂ (Ω)	R ₃ (Ω)	Q ₂ (F.s ^{a-1})	a ₂	Q ₃ (F.s ^{a-1})	a ₃
Control	(2)	25.8	119.6	13921	7.4x10 ⁻⁵	0.45	1.2x10 ⁻⁴	0.57
Electrochemical	(2)	23.5	80.3	1205	8.7x10 ⁻⁶	0.73	1.12x10 ⁻³	0.612

Table S2. Fitted parameters to EC 2 (control and electrochemical-treated electrodes).

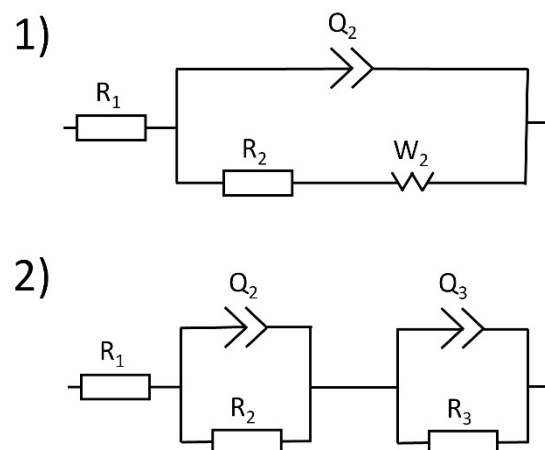


Figure S2. Equivalent circuits for abiotic electrode essays' modellization. 1) DMF and acetone 2) Control and electrochemical treatment.